

ASYMMETRIC DIELS-ALDER REACTION
PROMOTED BY A CHIRAL TITANIUM REAGENTKoichi NARASAKA, Masayuki INOUE, and Naoko OKADA
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The asymmetric Diels-Alder reaction between prochiral dienes and dienophiles prepared from α,β -unsaturated acids and 1,3-oxazolidin-2-one proceeds by the use of a chiral alkoxy titanium(IV) to give the corresponding cycloadducts in high enantioselectivity.

Due to the preeminent utility of the Diels-Alder reaction in organic synthesis, considerable attention has been given to the exploration of intermolecular asymmetric Diels-Alder reactions. Recent progress in this area has led to the development of various highly selective asymmetric Diels-Alder reactions by employing chiral dienes or dienophiles.¹⁾ On the other hand, little work has been done on asymmetric Diels-Alder reactions promoted by chiral Lewis acids²⁾ and this area remains as an unexplored field.

In this communication, we describe a highly enantioselective Diels-Alder reaction between prochiral dienes and dienophiles utilizing a chiral alkoxy titanium(IV).

Firstly, α,β -unsaturated acids of dienophiles were converted to the corresponding 3-acyl-1,3-oxazolidin-2-ones 1 based on the consideration that such bidentate dienophiles would form rigid complexes with a chiral Lewis acid to result in a good level of π -facial selectivity during the cycloaddition process. As a chiral Lewis acid, cyclic dialkoxy dichloro titanium(IV) 2 were chosen and prepared in situ from various chiral 1,2- or 1,4-diols 3 and dichlorodiisopropoxy-titanium(IV) according to the alkoxy exchange method.³⁾

The reaction of 3-crotonoyl-1,3-oxazolidin-2-one (1a) and cyclopentadiene was examined in toluene in the presence of an equimolar amount of various chiral alkoxy titaniums 2 (Table 1). When chiral 1,1,4,4-tetraphenylbutanetetraol derivatives 3b³⁾ and 3c were used as chiral auxiliaries, the endo adduct 4 was obtained in good enantioselectivity. Especially, the titanium reagent derived from 2,3-O-phenylethylidene derivative 3c⁴⁾ was found to promote the asymmetric cycloaddition reaction in a selective manner, and the optical purity of the adduct 4a reached up to 75% ee. Moreover, by the use of 2 molar amounts of the titanium reagent 2c, the product 4a was obtained in 92% ee.

Since a high level of enantioselectivity was achieved by the employment of the chiral titanium reagent 2c, the asymmetric Diels-Alder reaction of various 1,3-oxazolidin-2-one derivatives of α,β -unsaturated acids and cyclopentadiene was studied in the presence of 2 molar amounts of the titanium reagent 2c and the results are listed in Table 2. Excepting the acryloyl derivative 1b, various dienophiles 1 were found to react with cyclopentadiene in a highly selective manner to afford the corresponding endo adducts 4 in high optical purity (80-92% ee).

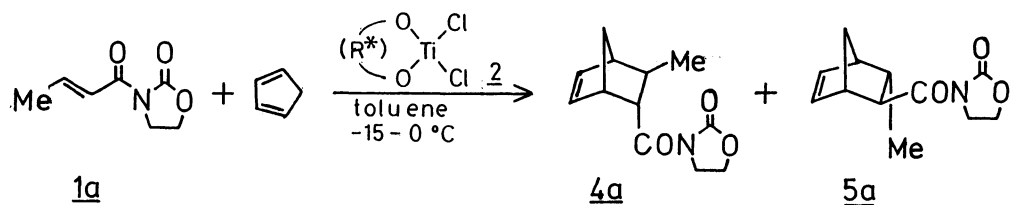
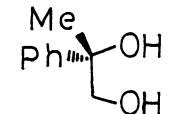
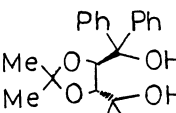
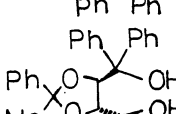
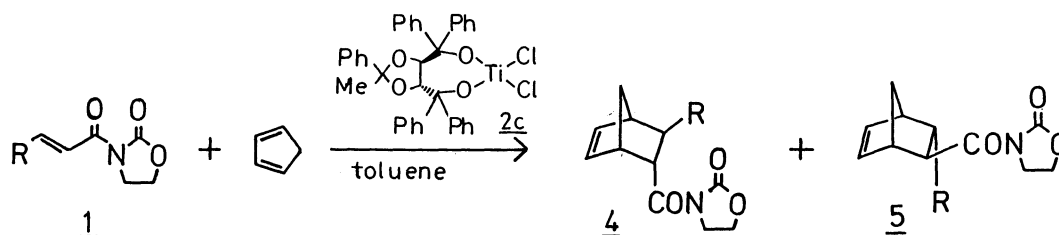


Table 1. Effect of Chiral Diols 3 in the Reaction of 1a with Cyclopentadiene

Diol <u>3</u>	Yield/% ^{a)}	endo:exo (<u>4a</u> : <u>5a</u>)	ee/% of <u>4</u> ^{b)}	Abs.config.
	<u>3a</u> 73	86 : 14	2	2R, 3S
	<u>3b</u> 88	93 : 7	55	2S, 3R
	<u>3c</u> quant.	86 : 14	75	2S, 3R
	93 ^{c)}	90 : 10	92	2S, 3R

Molar ratio of dienophile : titanium reagent = 1 : 1.

- a) Total yield of the endo and exo isomers which were separated by column chromatography on silica gel.
- b) The product was esterified to the benzyl ester by Evans's procedure,⁵⁾ and the optical purity was determined by the optical rotation.^{1d)}
- c) Molar ratio of dienophile : titanium reagent = 1 : 2.

Table 2. Asymmetric Diels-Alder Reaction of 1 with Cyclopentadiene

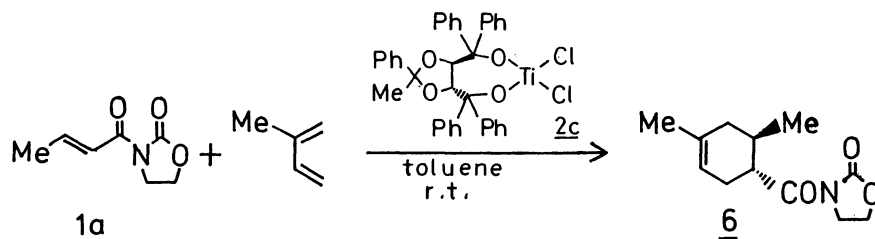
Dienophiles R	React.temp / °C	Yield/% ^{a)}	endo:exo	ee/% of <u>4</u>	[α] _D /°	Abs.config.
Me <u>1a</u>	-15	93	90 : 10	92 ^{b)}	-191 (c 3.6, CCl ₄)	2S, 3R
H <u>1b</u>	-78	69	86 : 14	38 ^{b)}	-65 (c 1.5, CHCl ₃)	2S
Ph <u>1c</u>	0	97	92 : 8	81 ^{c)}	-143 (c 1.2, CCl ₄)	
n-Pr <u>1d</u>	-15	82	90 : 10	90 ^{d)}	-158 (c 2.0, CCl ₄)	
Me <u>1e</u>	r.t.	77	92 : 8	82 ^{d)}	-142 (c 3.5, CCl ₄)	

Molar ratio of dienophile : titanium reagent = 1 : 2.

All the products gave satisfactory ¹H NMR and IR spectra.

- a) Total yield of endo and exo isomers. These isomers were separated by column chromatography on silica gel.
- b) The products were converted to the corresponding benzyl esters by Evans's procedure,⁵⁾ and the absolute configuration and the optical purity were determined by the optical rotation.^{1d)}
- c) The product was reduced to an alcohol with lithium aluminum hydride, and the optical purity was determined by ¹⁹F NMR analysis of the corresponding chiral MTPA ester.⁶⁾
- d) The products were reduced to alcohols with lithium aluminum hydride, and the optical purity was determined by HPLC analysis of the corresponding chiral Pirkle's carbamates.⁷⁾ (Waters μPolasil, hexane : ethyl acetate = 30 : 1)

Furthermore, high enantioselectivity was observed by the application of the chiral titanium reagent 2c to the reaction of 1a and an acyclic diene such as isoprene, and the cycloadduct 6 (the absolute configuration was not determined) was obtained in 92% ee.



Thus, according to the present methods utilizing the chiral alkoxy titanium(IV), it is noted that high enantioselectivity is achieved in the asymmetric Diels-Alder reaction between prochiral dienes and dienophiles.

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References

- 1) a) L. A. Paquette, "Asymmetric Synthesis," ed by J. D. Morrison, Academic Press, Inc., New York (1984), Vol.3, Chap.4; b) W. Oppolzer, *Angew. Chem., Int. Ed. Engl.*, **23**, 876 (1984); c) W. Oppolzer, C. Chapuis, and G. Bernardinelli, *Helv. Chim. Acta*, **67**, 1397 (1984); d) D. A. Evans, K. T. Chapman, and J. Bisaha, *J. Am. Chem. Soc.*, **106**, 4261 (1984); e) B. M. Trost, D. O'Krongly, and J. L. Belletire, *J. Am. Chem. Soc.*, **102**, 7595 (1980).
- 2) S. Hashimoto, N. Komeshima, and K. Koga, *J. Chem. Soc., Chem. Commun.*, **1979**, 437; M. Bednarski, C. Maring, and S. Danishefsky, *Tetrahedron Lett.*, **24**, 3451 (1983).
- 3) D. Seebach, "Modern Synthetic Methods," ed by R. Scheffold, John Wiley & Sons, Inc., New York (1983), Vol.3, Chap.4.
- 4) The diol **3c** was prepared from L-(+)-dimethyl tartrate. The reaction of the tartrate with 1,1-dimethoxy-1-phenylethane and a cat. amount of p-toluenesulfonic acid gave the 2,3-O-phenylethylidene derivative, which was converted to the diol **3c** by treatment with excess phenylmagnesium bromide. The diol **3c** was purified by column chromatography on silica gel (hexane : ethyl acetate = 5 : 1), and recrystallization of **3c** from a mixture of hexane and 2-propanol gave the complex of **3c** with 2-propanol as a white crystal (mp 111-114 °C). Then, by the azeotropic removal of 2-propanol with benzene was obtained the diol **3c** as a white amorphous solid. The enantiomeric isomer of the diol **3c** was not observed by HPLC analysis (Chiralpak OT(+), Daicel Chemical Industries; hexane : 2-propanol = 7 : 1).
- 5) D. A. Evans, M. D. Ennis, and D. J. Mathre, *J. Am. Chem. Soc.*, **104**, 1737 (1982).
- 6) J. A. Dale, D. L. Dull, and H. S. Mosher, *J. Org. Chem.*, **34**, 2543 (1969).
- 7) W. H. Pirkle and M. S. Hoekstra, *J. Org. Chem.*, **39**, 3904 (1974).

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